

AN INTRAMOLECULAR [8+6] CYCLOADDITION

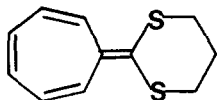
Ching-Yang Liu, Douglas A. Smith, and K.N. Houk*¹

Department of Chemistry, University of Pittsburgh
Pittsburgh, Pennsylvania 15260

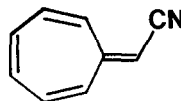
Abstract: An intramolecular [8+6] cycloaddition of a heptafulvene linked to a fulvene by a trimethylene chain has been observed.

Interest in the discovery of cycloadditions involving more than 6π electrons has been sustained for the twenty years since the creation of the Woodward-Hoffmann selection rules for concerted cycloadditions.² An [8+2] cycloaddition of heptafulvene was discovered by Doering and Wiley in 1960,³ and many other examples of [8+2] cycloadditions involving substituted heptafulvenes have been found.⁴ A variety of [6+4] cycloadditions have been discovered.⁵ Prinzbach has reported several [12+2] cycloadditions.⁶ A [$\pi_{14a} + \pi_{2s}$] cycloaddition involving heptafulvene and TCNE was discovered by Doering,⁷ and has become rather legendary in this field. Since suprafacial cycloadditions involving $(4n+2)\pi$ electrons can occur in a concerted fashion, and because we had experience with both intramolecular heptafulvene (8π)⁴ and fulvene (6π)⁸ cycloadditions, we undertook a search for an [8+6] cycloaddition. An intramolecular 14π electron [8+6] cycloaddition reaction is reported here.

Since dimethylfulvene undergoes concerted cycloadditions to electron-rich dienes exclusively as a 6π addend, we first attempted an intermolecular [8+6] cycloaddition of dimethylfulvene to the electron-rich 8,8-dithioheptafulvene, **1**.⁹ Heating a 1:1 mixture of these in refluxing xylene failed to give any [8+6] adduct, but gave only polymerization of the addends.



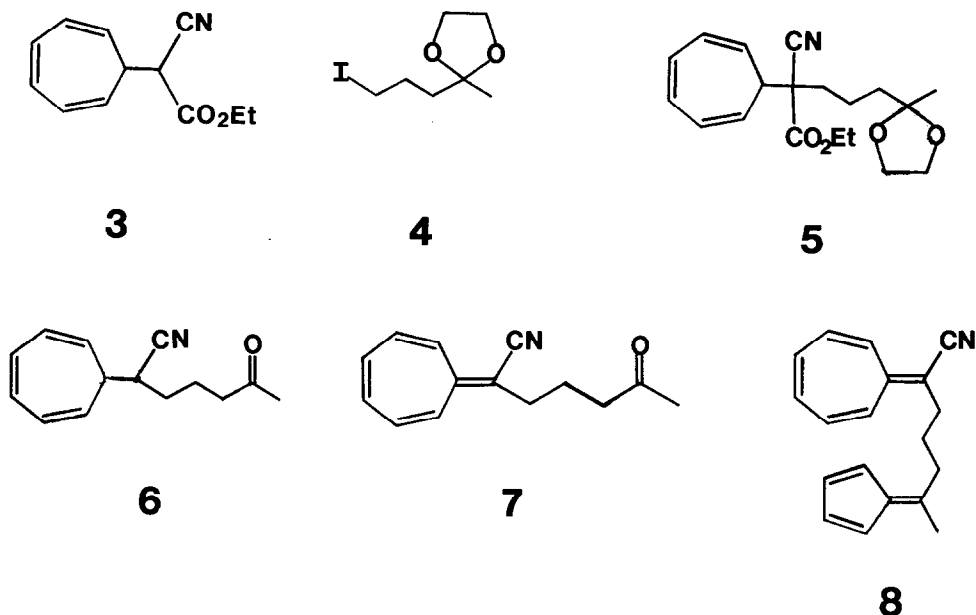
1



2

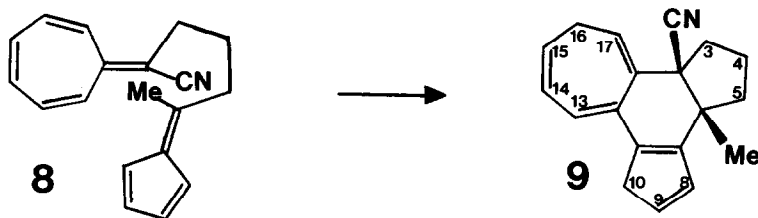
Similarly, no reaction was observed with this heptafulvene and the electron-rich 6-(dimethylamino)fulvene. The attempted reaction of 8-cyanoheptafulvene, **2**, with dimethylfulvene or 6-(dimethylamino)fulvene in refluxing xylene yielded complex reaction mixtures and decomposition. No products of these reactions have been identified. Since intermolecular [8+6] cycloadditions failed, a fulvene connected to a heptafulvene was synthesized, in order to attempt an intramolecular cycloaddition.

The heptafulvene-fulvene, **8**, was prepared from the intermediates shown below.



Cycloheptatriene **5** was obtained in 78% yield by treating ethyl tropyloxyacetate, **3**, with sodium ethoxide and 5-iodo-2-pentanone ethylene ketal, **4**,¹⁰ in dry ethanol. Hydrolysis and decarboxylation of **5** gave **6** in 46% yield after column chromatography. Thermal isomerization of **6**, followed by treatment with trityl fluoroborate and triethylamine¹¹ gave heptafulvene, **7** in 31% yield, which was converted to **8**¹² in 81% yield by reaction with cyclopentadiene and methylamine in dry methanol.¹³

Thermolysis of **8** in toluene at 185°C for 2 h in a sealed tube in the presence of BHT gave **9** in 48% yield, along with 32% yield of a mixture of compounds. The major product, **9**,



proved to be the result of an intramolecular [8+6] cycloaddition, followed by 1,5-sigmatropic shifts in both the cycloheptatriene and cyclopentadiene moieties.

The structure of **9** was confirmed by analysis of the 300 MHz ¹H NMR spectrum. A sharp singlet at δ 1.32 is assigned to the methyl group. A six-proton signal at δ 1.75-2.29 is assigned to H-3-H-5. The doublet at δ 6.59 ($J_{13,14} = 5.5$ Hz) is assigned to H-13. On irradiation at δ 6.17, the multiplet at δ 5.57 collapsed to a triplet ($J_{15,16} = 7$ Hz) and the doublet at δ 6.59 ($J_{13,14} = 5.5$ Hz) collapsed to a sharp singlet. Therefore, one-proton signals at δ 6.17 and δ 5.57 are assigned to H-14 and H-15, respectively. On irradiation at

δ 5.57 (H-15), the multiplet at δ 6.17 (H-14) collapsed to a doublet ($J_{13,14} = 5.5$ Hz), the multiplet at δ 2.82 collapsed to a doublet of doublets ($J_{\text{gem}} = 13$ Hz, $J_{16a,17} = 8$ Hz), and the multiplet at δ 2.0 is simplified. On irradiation at δ 2.82, the multiplet at δ 5.57 (H-15) collapses to a doublet of doublets ($J_{14,15} = 9$ Hz, $J_{15,16b} = 7$ Hz), the doublet of doublets at δ 5.97 is collapsed to a doublet ($J_{16b,17} = 6$ Hz), and the multiplet at δ 2.0 is simplified. Therefore, the multiplet at δ 2.82 ($J_{16a,17} = 7$ Hz), the multiplet at δ 2.0, and the doublet of doublets at δ 5.97 ($J_{16a,17} = 8$ Hz, $J_{16a,17} = 6$ Hz) are assigned to H-16a, H-16b and H-17 respectively. The doublets at δ 3.11 ($J_{\text{gem}} = 22$ Hz, $J_{9,10} = 1$ Hz) and δ 3.35 ($J_{\text{gem}} = 22$ Hz, $J_{9,10} = 1$ Hz) are assigned to H-10. The doublet of triplets at δ 6.48 ($J_{8,9} = 5.5$ Hz, $J_{8,10} = 1$ Hz) and δ 6.41 ($J_{8,9} = 5.5$ Hz, $J_{9,10} = 1$ Hz) are assigned to H-8 and H-9, respectively.

The cis stereochemistry of **9** was inferred from the relatively low-field position of the quaternary methyl resonance, which indicates that the methyl group resides in the deshielding zone of the vicinal group.¹⁴

Four double bond isomers have structures which are compatible with nmr spectral data given above. A series of NOE experiments at 600 MHz led to the assignment of regiochemistry. Upon irradiation of the methyl singlet at δ 1.32, a large positive enhancement at δ 6.48 (H-8) and a small positive enhancement at δ 5.97 (H-17) were observed. No enhancement was observed at δ 3.11 (H-10), 3.35 (H-10) or for the rest of the cycloheptatriene ring system. These results are consistent with the regiochemistry of the cyclopentadiene and cycloheptatriene units shown in structure **9**. This was supported by the positive enhancement at δ 2.19 (H-5) and δ 1.32 (methyl), upon irradiation at δ 6.48 (H-8). A small positive enhancement at δ 5.97 (H-17) was also observed, which seems surprising in light of the larger distance between H-8 and H-17. Upon irradiation at δ 6.59 (H-13), only positive enhancement at δ 3.11 (H-10) and δ 3.35 (H-10) was observed. Irradiation at δ 5.97 (H-17) produces a negative enhancement at both δ 6.48 (H-8) and δ 1.32 (methyl), confirming the results stated above. However, no enhancement was observed between δ 1.75-2.29 (H-3-H-5). This suggests the presence of a cis-fused ring junction, in which the cyano group, methyl group, H-17 and H-8 are all above the six-membered ring, while the saturated five-membered ring is down. On the basis of these data, structure **9** is proposed for the major adduct. The minor product of the thermolysis appears, from NMR coupling patterns and mass spectral data, to be a mixture of intramolecular [8+6] cycloaddition products in which the positions of the five- and seven-membered ring methylenes differ.

Prinzbach¹⁵ has made a similar benzenoid dihydrocyclohept[e]indene via thermal 14π -electron electrocyclization of sesquifulvalene. Both regioisomers arising from [1,5]-sigmatropic hydrogen shifts in the five-membered ring were observed and were shown to be non-interconvertible.

Acknowledgement. We are grateful to the National Institutes of Health for financial support of this research. The 600 MHz nmr spectra were obtained at the Carnegie-Mellon University Facility supported by the National Institutes of Health.

References and Notes

1. Address correspondence to Department of Chemistry and Biochemistry, University of California, Los Angeles, California 90024.
2. (a) R.B. Woodward and R. Hoffmann, "The Conservation of Orbital Symmetry," Verlag Chemie, Weinheim, 1970; (b) C.-Y. Liu, J. Mareday, K.N. Houk, and F.R. Fronczek, J. Am. Chem. Soc., **105**, 6714 (1983).
3. W. von E. Doering and D.W. Wiley, Tetrahedron, **11**, 183 (1960).
4. C.-Y. Liu, J. Mareday, and K.N. Houk, J. Am. Chem. Soc., **105**, 6714 (1983).
5. R.C. Cookson, B.V. Drake, J. Hudec, and A. Morrison, Chem. Commun., **15** (1966); T. Nozoe, T. Mukai, K. Takase, and T. Takase, Proc. Japan Acad., **28**, 477 (1952); M.E. Garst, V.A. Roberts, K.N. Houk, and N.G. Rondan, J. Am. Chem. Soc., **106**, 3382 (1984), and references therein.
6. H. Prinzbach and H. Knofel, Angew. Chem. Int. Ed. Engl., **8**, 881 (1969); A. Beck, D. Hunkler, and H. Prinzbach, Tetrahedron Lett., **24**, 2151 (1983).
7. W. von E. Doering, personal communication to R.B. Woodward, reported in reference 2(a).
8. Y.N. Gupta, M.J. Doa, and K.N. Houk, J. Am. Chem. Soc., **104**, 7336 (1982); T.-C. Wu, J. Mareday, Y.N. Gupta, and K.N. Houk, Ibid. **105**, 6996 (1983).
9. K.M. Rapp and J. Daub, Tetrahedron Lett., 2011 (1976).
10. R.D. Miller and D.R. McKean, J. Org. Chem., **46**, 2412 (1981).
11. M. Oda and Y. Kitahara, Chem. Commun., 352 (1969).
12. The 300 MHz nmr spectrum of **8** showed the following: δ 6.8 (bdm 1H, $J_{1,2} = 11$ Hz); δ 6.5 (bm, 4H); δ 6.2 (bm, 5H); δ 2.59 (t, 2H, $J_{9,10} = 7.5$ Hz); δ 2.23 (s, 3H); δ 2.21 (t, 2H, $J_{10,11} = 7.5$ Hz); δ 1.82 (quint, 2H, $J_{9,10} = J_{10,11} = 7.5$ Hz).
13. R.D. Little and G.W. Muller, J. Am. Chem. Soc., **101**, 7129 (1979).
14. M. Onda, M. Gotoh, and Okada, J. Chem. Pharm. Bull., **23**, 1561 (1975).
15. H. Prinzbach, H. Bingmann, A. Beck, D. Hunkler, H. Sauter, and E. Hadicke, Chem. Ber., **114**, 1697 (1981).

(Received in USA 7 July 1986)